

Science 3 – 5 for use with NGSS 2013:

Physical Science: Matter and Its Interactions

Conservation of Mass and Matter

Students	Learning Continuum Statements:
----------	--------------------------------

Students:

RIT 181-190:

- Applies the law of conservation of matter to calculate mass/weight of parts of mixtures

Students:

RIT 191-200:

- Applies the law of conservation of matter to calculate mass/weight after a chemical change
- Applies the law of conservation of matter to determine mass/weight after changes of state
- Applies the law of conservation of matter to calculate mass/weight of parts of mixtures

Students:

RIT 201-210:

- Applies the law of conservation of matter to calculate mass/weight after a chemical change
- Applies the law of conservation of matter to determine mass/weight after changes of state
- Determines evidence that supports a claim that substances contain matter
- Applies the law of conservation of matter to explain the results of investigations involving chemical reactions
- Applies conservation of matter to explain results of investigations involving physical changes
- Applies the law of conservation of matter to calculate mass/weight of parts of mixtures
- Relates changes in temperature and state to changes in thermal energy
- Plans investigations to provide evidence of conservation of matter

Students:

RIT 211-220:

- Applies the law of conservation of matter to calculate mass/weight after a chemical change
- Applies the law of conservation of matter to determine mass/weight after changes of state
- Recognizes how open containers affect changes in mass during physical changes
- Applies the law of conservation of matter to explain the results of investigations

involving chemical reactions

- Applies the law of conservation of matter to calculate mass/weight of parts of mixtures
- Relates changes in temperature and state to changes in thermal energy
- Plans investigations to provide evidence of conservation of matter

Students:

RIT 221-230:

- Applies the law of conservation of matter to infer whether systems are open or closed
- Applies the law of conservation of matter to calculate mass/weight after a chemical change
- Applies the law of conservation of matter to determine mass/weight after changes of state
- Applies the law of conservation of matter to explain the results of investigations involving chemical reactions
- Relates changes in temperature and state to changes in thermal energy
- Plans investigations to provide evidence of conservation of matter

